

Interpreting Solubility Curves Pogil Answerkey

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Interpreting Solubility Curves 5. Use the temperature dependence of solubility to identify whether the substances in Table G are gases or solids. Make two lists below, one for gases and one for solids. 6. Suggest a reason why solubility decreases with increasing temperature for gaseous solutes but increases for solid solutes. fum and Problems 1.

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Interpreting Solubility Curves Model Table1. Solubility Data Temperature (°C) Solute (g) per 100g of H₂O (l) 10 33 30 42 50 52 70 62 90 73 Task Complete the model: Using the grid below make a graph of the solubility data in Table 1. a. Label x-axis, y-axis and create appropriate scales for each. b. Plot points using a pencil Key Questions 1.

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Solubility Curves. A series of free IGCSE Chemistry Activities and Experiments (Cambridge IGCSE Chemistry). Saturated solution - When a maximum amount of the solute is dissolved. Unsaturated solution - When less than a maximum amount of the solute is dissolved. When saturation has been reached - Rate of dissolving = Rate of crystallisation.

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SOLLUBILITY CURVE WORKSHEET KEY Use your solubility curve graphs provided to answer the following questions. 1. What are the customary units of solubility on solubility curves? Degress Celsius and grams of solute/100g of water 2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent. 3.

SOLUBILITY CURVE WORKSHEET

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